(1980) Leaching Behavior of rhyolite glass. Nature, v.284, no.5755, pp.435-437.

- KARKHANIS S.N., MELLING, P.J., FYFE, W.S. and BANCROFT, G.M. (1981) Stable Product Low Leach Glasses: *In:* John G. Moore (Ed.), Scientific Basis for Nuclear Waste Management, v.3, Plenum Publishing Corporation, 1981, pp.115-122.
- MELLING P.J., KARKHANIS, S.N. FYFE, W.S. and BANCROFT, G.M. (1983), The hydrothermal leaching behavior and properties of sodium-calcium aluminophosphosilicate glasses. Glass Technology, v.24, no.4, pp.192-197.
- NESBITT, H.W., BANCROFT, G.M., FYFE, W.S., KARKHANIS, S.N., NISHIJIMA, AKIO and SHIGEMITSU SHIN. (1981a) Thermodynamic stability and kinetics of peroveskite dissolution, Nature, v.289, no.5796, pp.358-362.
- NESBITT, H.W., BANCROFT, G.M., FYFE, W.S., KARKHANIS, S.N., MELLING, P.J. and Nishijima, A. (1981b) On the thermodynamic stability and kinetic dissolution of perovskite in natural waters. Alternate Nuclear Waste forms and interactions in geology.

Proceeding of Workshop Summarizing Advanced Activities and Formulating for the Future-May 13-15,Gatlinburg, Tennessee-Cosponsored by Oak Ridge National Laboratory and U.S. Department of Energy, Division of Materials Sciences.

- NESBITT, H.W., BANCROFT, G.M., KARKHANIS, S.N. and FYFE, W.S. (1980) The stability of Titanium minerals in presence of backfill & repository materials A General Approach. Symp. D "Scientific Basis for Nuclear Waste Management-November 17-20, 1980, Materials Research Society Annual Meeting Nov.16-21, 1980, Boston, Massachusetts.
- RINGWOOD, A.E., KESSON, S.E., WARE, N.G., HIBBERSON, W. and MAJOR A. (1979a) Immobilisation of high-level nuclear reactor wastes in SYNROC. Nature, v.278, pp.219-223.
- RINGWOOD, A.E., KESSON, S.E., WARE, N.G., HIBBERSON, W. and MAJOR A. (1979b) The Synroc Process: A Geochemical approach to Nuclear Waste Immobilization Geochemical Jour., v.13, pp141-165.

NEW OCCURRENCE OF MANGANOCOLUMBITE FROM LATE PROTEROZOIC PEGMATITES OF BHURPIDUNGRI, JHARSUGUDA DISTRICT, ORISSA by P. Jagadeesan, K.S. Mishra, and P.V. Ramesh Babu. Jour.Geol. Soc. India, v.66, 2005, pp.141-144.

S. Viswanathan, Flat B-203, Block-B, United Avenue Apartments, South End, 7-1-29, Ameerpet, Hyderabad-500016, comments:

The authors claim that they are reporting for the first time, the occurrence of manganocolumbite in the pegmatites of India at Bhurpidungri in the Jharsuguda district of Orissa. This claim cannot be accepted for the following reasons: Two criteria have to be fulfilled for naming a member of the columbite-tantalite isomorphous series as 'mangano-columbite': (1) an Mno/FeO ratio of more than 3 and (2) a Ta₂O₅ content of less than 20 percent by weight (Vlasov, 1966, p.453; Kuz'menko, 1959, in Vlasov, *op.cit*). Kuz'menko and Vlasov also feel that the term 'mangano-columbite' should be reserved only for the manganese end-member of the columbite-tantalite series, MnNb₂O₆.

In Table 2 on p.143, the authors have given analytical data for eight samples of columbite-tantalite, numbered BP/1 to BP/8, from Bhurpidungri. As they have not given the MnO/FeO ratio of the samples, and also overlooked the significance of these ratios, I have calculated the MnO/FeO ratios. Seven of the samples (BP/1, 2, 4 to 8) have MnO/FeO ratios of less than 3, with BP/5 having a very low value of 0.86. Although one sample (BP/3) has an MnO/FeO ratio of

3, its Ta_2O_5 content is high (24.45%). The average of the eight samples (BP_x) has an MnO/FeO ratio of only 2, though its Ta_2O_5 content is 14.33%. Therefore, the Bhurpidungri columbite-tantalites are not 'manganocolumbites'.

On p.144, the authors state that, "The manganocolumbite under study contains up to 18.46% MnO, and is comparable with that from the San Diego Mine, Mesa Grande, California (MnO - 19.21%) and Pakeagama pegmatite, Ontario, Canada (MnO - 12.50%) (Breaks et al. 1998) (Table 3)." This comparison is misleading because, when comparing 'manganocolumbites' from different areas, it is not enough to compare only their MnO values. It is absolutely essential to compare their MnO/FeO ratios and Ta2O5 contents. For instance, with its very high MnO/FeO ratio of 55.06 and Ta₂O₅ of 12.56%, the San Diego Mine sample is a fine example of a manganocolumbite. In sharp contrast, with its low MnO/FeO ratio of 2.17 and high Ta₂O₅ of 30.99%, the Pakeagama sample is not a manganocolumbite. As already pointed out, the 'average' Bhurpidungri sample is also not a manganocolumbite because of its low MnO/FeO ratio of 2.

If the MnO content of a columbite-tantalite is the only criterion for naming it as 'manganocolumbite', as the authors appear to believe, they should note the following: Columbite-tantalites with high MnO contents, within the range of 8 91 to 18 46% given by them for the so-called 'manganocolumbites' of Bhurpidungri in Table 2 (p 143), occur at Faguni, Gadar, Tharipahari, Saknakola, Kararua, Ambadah, Kalapahari, Tarazoppa, Goriadih, Nirupahari, and Domchanch in the Bihar mica-pegmatite belt (B N Tikoo, 1990 The geochemistry of columbite-tantalite from the raremetal pegmatite belts of India Unsubmitted Ph D Thesis He died before he could submit his Thesis to the Osmania University I was his Ph D Guide) The columbite-tantalite from Tarazoppa (Hazarıbagh district, 72 H/10, 24°31'20" $85^{\circ}44'40''$) (MnO = 12 93%, FeO = 4 03%, Nb₂O₅ = 49 40%, Ta₂O₅= 31 09%) is strikingly similar to the one from the Pakeagama pegmatite, Ontario, Canada (MnO = 1259%, FeO = 562%, Nb₂0₅ = 4882%, Ta₂O₅ = 3099%), which the authors consider as being a 'manganocolumbite', citing Breaks et al 1998, but is not, because of its low MnO/FeO ratio of only 2 17 and high Ta₂O₅ (30 99%) Despite its much higher MnO/FeO ratio of 3 21, the Tarazoppa columbite-tantalite is also not a 'manganocolumbite' because of its high Ta₂O₅ (31 09%)

On p 143, the authors have stated that "two samples were investigated by the X-ray diffraction method that confirmed it as manganocolumbite (PDF data Card No 33-899 of JCPDS)" They refer to the two samples as "Sample-I" and "Sample-II" Are these two from the batch of eight, numbered BP/1 to BP/8, and if so, which two? If not, they should have given analytical data for these two samples also, with the MnO/FeO ratio, Ta_2O_5 content, and locality of the manganocolumbite listed in PDP Data Card No 33-899 of JCPDS

If the nomenclature of different members of the columbite-tantalite isomorphous series has undergone any revision in recent years, I wish to get enlightened by the authors

P. Jagadeesan¹, K.S. Misra¹ and P.V. Ramesh Babu², Atomic Minerals Directorate for Exploration and Research, ¹Civil Lines, Nagpur – 440 001, ²Begumpet, Hyderabad – 500 016, reply

The authors are sincerely thankful to Dr S Viswanathan

for his valuable comments We would like to make the following observations

- 1 The first claim of manganocolumbite is genuine and so far none has reported manganocolumbite from Indian pegmatites The examples quoted by Dr S Viswanathan that some of the pegmatites viz, Faguni, Gadar, Tharipahari etc, in Bihar Mica Belt having high content of MnO have not been published so far
- 2 The Ta_2O_5 content has nothing to do with the nomenclature of manganocolumbite, instead, if it is high >50%, it will be known as manganotantalite Vlasov (1966) has also advocated that the ratio of Fe/ Mn can be used for prefixing "iron" or "manganese" in front of the mineral or by using Mn-tantalite, Mncolumbite etc, (Vlasov, 1966, p 453) Presently naming of the mineral has been preferred by taking elemental ratio of Mn/Mn+Fe (atomic ratio) (after Cerny et al 2003) instead of taking MnO/FeO ratios
- 3 His comment on percentage of Ta_20_5 that it should be <20% is not correct. The manganocolumbite mineral data clearly shows that it can be >20% and can go up to 33 58% (Cerny et al. 2003, Breaks et al. 1999)
- 4 Regarding his comment on the comparison of manganocolumbite of Bhurpidungri with Pakeagama sample, it is clarified that it is well-published data by Breaks et al (1999) in which they have clearly mentioned that it is a manganocolumbite and comparison is not only done for its percentage of MnO₂ but considering all the major oxides and its geological set up
- 5 The sample Nos I and II are from the same batch and they are sample Nos BP/2 and BP/3
- 6 We agree with Dr S Viswanathan that average may not be a representation of manganocolumbite because there are few samples which may not be true manganocolumbite (Sample Nos BP/5 & BP/8) but in a pegmatite there will not be a single mineralogical representation but they occur in a isomorphous solid solution series. This is the reason for variation in their chemical composition.

References

BREAKS, FW, TINDLE, A G and SMITH, S R (1998) Rare metal mineralisation associated with the Berens River-Sachigo subprovincial boundary, northwestern Ontario discovery of a new zone of complex type, petalite subtype pegmatite and implications for future exploration In Ontario Geol Surv Misc Paper 169, pp 168-182

BREAKS, FW, TINDLE, A G and SMITH, S R (1999) The Pakeagama lake Pegmatite Continued Field and laboratory Investigation

of Highly Evolved, Complex Type, Petalite Subtype Rare – Element Mineralization in the Berens River, Sachigo Subprovince Boundary Zone (website www.houston lakemining.com/reports/report1.html.)

CERNY, P, CHAPMAN, R, FERREIRA, K and SMEDS, S A (2003) Geochemistry of Nb-Ta-Oxide Minerals in the Varutrask Pegmatite The Case of a "Misplaced" Columbite-to-Tantalite Trend Dept Geol Sci, Univ of Manitoba (website umanitoba ca/faculties/ science/ /grads/colloquium/2002-2003/cerny htm)

- KUZ'MENKO, M V (1959) K geokhimi tantala l niobiye (The geochemistry of tantalum and niobium) Trudy IMGRE AN SSR No 3, pp 3-25
- VKASOV, K A (1966) Geochemistry and mineralogy of rare elements and genetic types of their deposits Vol 2 Mineralogy of rare elements Israel Program for Scientific Translations, Jerusalem, p 453

GEOSTATISTICAL STUDIES OF A GOLD PROSPECT IN SIDHI DISTRICT, MADHYA PRADESH by K. Saikia and B C Sarkar. Jour. Geol Soc. India, v 66, pp.229-241

Shakeel Ahmed, National Geophysical Research Institute, Hyderabad, Email: shakeelahmed@ngri res in, comments

The authors of the article deserve appreciation for applying geostatistics, to the estimation of a gold deposit and publishing the same

The approach in this article is similar to the one published by Sarkar and Roy (2005) Also following two points are required to be explained

- 1. When they fitted log normal model to gold distribution, why did they not carry log-normal kriging This point needs clarification and elaboration
- 2 My comments in respect of grade tonnage curves, given in my comments on Sarkar and Roy (2005, please refer p 542 of this issue) also hold good in respect of this paper. This points need clarification
- K. Saikia and B.C. Sarkar, Indian School of Mines, Dhanbad, *Email: bhabesh_sarkar@yahoo co in, reply*

Our point wise replies to the comments are as follows

 Reasons for applying ordinary kriging for estimation of block values instead of lognormal kriging are mentioned in the paper on p 237 under the section 'Block Kriging', lines 8 to 11 References of Champigny and Armstrong (1988) and David (1968) clearly state the reasons For the sake of clarification, these are explicitly given below

David (1988) in his book 'Handbook of Applied Advanced Geostatistical Ore Reserve Estimation', p 49 states 'Circumstances of non-stationarity may render the estimation of the parameters of a log normal mode extremely difficult Hence, one may use the semi-variogram model that appears to be the best for deriving the parameters of a model'

Champigny and Armstrong (1988) in their paper on 'Estimation of Fault-controlled Deposit' and discussed on modeling of a gold deposit of epithermal type (3rd International Geostatistics Congress, Avignon, Sept 5-9, 1988, pp 311-322) They state 'Lognormal krigning was not considered in the modeling although values conformed to a lognormal distribution because it requires second order stationarity of $log_e(XJ)$ values and not just local stationarity'

- 2 The comments point to the single issue, i.e. deriving grade-tonnage curves Different authors have applied different methods that can be grouped as
 - Grade-tonnage curves derived from a histogram of sample grades
 - (II) Grade-tonnage curves derived from a continuous distribution representing sample grades;
 - (III) Grade tonnage curves derived from local block estimates,
 - (iv) Grade tonnage curves by multiple indicator kriging,
 - (v) Grade tonnage curves based on conditional block simulation

Virtually all these methods of deriving grade-tonnage curves contain some error. In the present modeling study, the grade-tonnage curves have been derived from local block estimates

The authors express deep sense of gratitude to Dr Shakeel Ahmed for going through the two of our above mentioned papers very minutely and objectively and offering his comments In fact, such an exercise aids in bringing improvements in the methodology