REE geochemistry of monazites from coastal sands between Bhimunipatnam and Konada, Andhra Pradesh, East coast of India

K. Bangaku Naidu*, K. S. N. Reddy, Ch. Ravi Sekhar, P. Ganapati Rao and K. N. Murali Krishna

Department of Geology, Andhra University, Visakhapatnam 530 003, India

The rare earth elements (REE) geochemistry of monazites of Bhimunipatnam-Konada coastal sand deposit was studied using EPMA method. The average LREE concentration was 53.31%, which is more than HREE (av 1.38%). Σ LREE more than actinides (Th + U) indicates that provenance for monazite in the study area is garnet-bearing paragenesis rocks such as charnockites and metapelitic rock (khondalite). The REE fractionation patterns and positive europium anomalies indicate that monazites were formed from magma/ anatectic melt with high oxygen fugacity. The U-Th-Pb geochemical dating of monazites is 1000 Ma (average), which indicates that they are derived from protoliths of charnockites and metapelitic rocks such as khondalites, which are formed during meso-neo-Proterozoic ages in the Eastern Ghats Granulite Belt.

Keywords: Coastal sand deposits, geochemical dating, khondalites, monazites, rare earth elements.

MONAZITE is one of the important radioactive minerals associated with placer mineral deposits in India. It contains the radioactive element thorium and is a storehouse of the rare earth elements (REE). REE are a group of 17 metallic elements consisting of 15 lanthanides (La, Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb and Lu) along with scandium and yttrium. REE are divided into two groups, i.e. lighter rare earth elements (LREE)-La to Sm, and heavier rare earth elements (HREE)-Gd to Lu. LREE + Sc is known as cerium group and HREE + Y as yttrium group. ThO₂ and REE contents of monazite are most important for its extraction for commercial purpose. REE concentration and distribution patterns are more significant compared to major and trace elemental abundance and distribution to understand the nature of the provenance of sediments^{1,2}. REE abundance and distribution patterns in sediments indicate that their source rock.

Many researchers³⁻¹⁹ have reported monazite occurrence and distribution from coastal sands in the present study area and in other parts of Andhra Pradesh $(AP)^{20-29}$ and Odisha coasts³⁰⁻³⁵. Few have made an attempt to study geochemistry of monazite in placer deposits of $AP^{2,24}$ and Odisha¹, while others³⁶⁻⁴⁸ have reported geochemistry of monazites from various litho units of the Eastern Ghats Group of rocks.

The present study deals with REE geochemistry of monazite and its provenance of Bhimunipatnam–Konada placer deposit in northern Andhra Pradesh coast.

The 25 km coastal stretch of the study area (83°23'E to 83°36'E long. and 17°51'N to 18°02'N lat.) extends from the Pedda Gedda in the south to the Champavathi River in the north. The Gosthani River joins the Bay of Bengal at Bhimunipatnam. These ephemeral rivers originate in the Eastern Ghats hill ranges and constitute the drainage system. These rivers carry huge amounts of sediment and debouch into the Bay of Bengal at Chepala Uppada, Bhimunipatnam and Konada. The study area has different geological and geomorphic features generated by the rivers, small creeks and dynamic seasonal winds. The average width of coastal sand deposit is 980 m and dunes have maximum thickness of 18 m. At the 4 km south of Bhimunipatnam, an area of 10 km² covered with red sands extending 1.5-2.5 km inland from the beach and length of 5 km along the coast and other red sediment deposit occurring near Dibbalapalem, covered an area of 3.5 km² and extended 2 km along the coast and 1.2 km in land from the coast. Figure 1 shows the location map of the study area.

The Eastern Ghats Granulite Belt along the east coast of India extends over 1000 km. Major rock units in this granulite terrain are khondalites exhibiting compositional heterogeneity from sector to sector, and charnockites³⁶ and lesser abundance of basic granulites, intrusive alkaline rocks, anorthosites⁴⁹, pyroxene granulites, syn–posttectonic granites⁴¹, quartzites and leptynites.

A total of 25 surficial sediment samples (20 from coastal sands and five from red sediments) were collected along the Bhimunipatnam–Konada coast. Initially these samples were thoroughly washed with distilled water for removing the salts, treated with 15% hydrogen peroxide (H₂O₂) to remove organic matter and then with 10% dilute HCl to remove the shell material, and finally subjected to +230 mesh wet sieving. This was followed by the dry sieving into +60 (>0.25 mm), -60 to +120 (0.25 to 0.125 mm) and -120 to +230 (0.125 to 0.062 mm) ASTM mesh size fractions. The samples were subjected to heavy mineral separation using bromoform.

Monazites were identified under binocular petrological microscope based on their optical properties. Twenty-five grains were picked for geochemical analysis from coastal sands of Bhimunipatnam–Konada and red sediments.

CAMECA SX-100 Electron Probe Micro Analyzer (EPMA), Geological Survey of India, Hyderabad was used for geochemical analysis of monazite samples.

Thin sections of polished monazite grains of coastal sands (25) were selected for U–Th–Pb chemical dating using EPMA. The operating conditions for the analyses

^{*}For correspondence. (e-mail: naidu0756@gmail.com)



Figure 1. Sample location map of the study area.

were 20 kV of high voltage and 200 nA beam current with the minimum possible beam diameter. It was assumed that the characteristic X-rays were generated from a volume represented on the surface by an area of ~2 μ m. Higher beam current was used for grains containing low thorium in order to have better statistics and lower errors. Calibrations were carried out at 20 kV and 200 nA while analysis was done at higher beam current (200–400 nA), PbM α was measured on Logarithmic Poisson Execution Time (LPET) and the peak counting time was 300 s with background measured on both sides. For uranium, the U(M β) line was used in order to avoid the interference of the Th(M β) line with a peak counting time of 200 s. U and Th were measured on PET crystal. Thorium M α peak was also counted for 200 s. REE and the X-ray lines measured included La (L α), Ce (L α), Nd (L α), Pr (L β), Sm(L α), Ho (L β), Dy (L α) and Gd (L β) on spectral analysis considerations arrived at using virtual EDS. Counting time at peak varied between 50 and 85 s for these elements. Y(L α) line was used for yttrium and it

was counted for 50s on peak. Interference of Y(Lc) on Pb(M α) and ThMc on Pb(M α) was corrected after measuring the interfering lines during calibration and thereafter applying the overlap correction. Other experimental conditions adopted here were similar to those reported in the literature^{50–52}. Table 1 gives the geochemical data of monazite in the study area.

Monazite (La, Ce, Nd, Th, PO₄) is a phosphate of LREE and contains lesser amount of yttrium and HREE. The chemical composition of monazites of Bhimunipatnam–Konada coastal sand deposit is as follows: ThO₂ content varies from 3.78% to 13.39% (av 9.42%), Y₂O₃ from 0.00% to 2.26% (av 0.36%), SiO₂ from 0.48% to 2.85% (av 0.85%), CaO from 0.7% to 1.76% (av 1.29%), and UO₂ varies from 0.03% to 0.42% (av 0.13%) (Table 1). Σ REE ranges from 43.47% to 67.78% (av 54.69%), Σ LREE from 42.90% to 64.08% (av 53.31%), while Σ HREE average is 1.38% and its varies from 0.57% to 3.70% in the study area (Table 2).

In general, ThO₂ content in monazite is 4-12% and cheralite, a variety of monazite contains up to 30% ThO₂ (ref. 53). The average of ThO₂ content in studied samples is 9.42%; these are monazites not cheralites.

Khondalites from the Eastern Ghats Granulites Belt show uniform thorium content³⁹, but large variation is observed in charnockites of the Eastern Ghats Group of rocks³⁸. From the khondalites suite of rocks, in monazites ThO₂ ranges from 6% to 10% and in monazite from charnockites, ThO₂ ranges between 9% and 10% (ref. 46). ThO₂ content in the 25 grains of monazites studied was <10% for 50% of the grains, while the remaining 50% grains showed >10% of ThO₂ content in the study area.

The heavy mineral assemblage of Bhimunipatnam– Konada placer mineral deposit with decreasing abundance is ilmenite, sillimanite, garnet, rutile, zircon, monazite and magnetite. The monazite grains are colourless to yellow, and are characterized by rounded to sub-rounded shape. Monazite grains show moderate relief, pits and etch marks.

The mineral assemblage of the study area is well correlated with mineralogical composition of the Eastern Ghats Group of rocks. The khondalite suite of rocks (garnet-quartz-sillimanite schist and gneiss and garnet ferous quartzite, leptynites, quartz granulites, calk silicate rocks and quartz-garnet-graphite-sillimanite schist) is a major lithological unit consisting of major minerals such as garnet \pm biotite \pm silimanite \pm cordorite \pm K-feldspar \pm plagioclase feldspar + quartz and minor minerals magnetite, ilmenite, rutile, sphene, apatite and monazite³⁹. The charnockite suite of rocks such as charnockites and enderbites, which consist of ortho and clino pyroxenes, garnet, biotite, magnetite, ilmenite, monazite, K-feldspar, plagioclase feldspar and quartz³⁸, and monazites of variable size (200-2000 µm) were noticed in charnockites and tiny (~50 μ m) inclusions in garnet and cordierite⁴⁶ from the khondalite rocks.

In the present study, chondrite normalized values of REE [lighter lanthanides (LLn) and heavier lanthanides (HLn)] were used to understand the provenance of monazites⁵⁴. REE Lu (0.85 Å)–La (1.06 Å) have higher charges and larger ionic radii and low concentration, showing little tendency to replace major elements during magmatic crystallization. Y and HREE have smaller ionic radii than LREE, due to which yttrium and HREE show low affinity for monazite crystal structure⁵⁵.

The ratio of $(\Sigma LLn/HLn)_{cn}$ in monazite of the coastal sands of the study area varies from 4.13 to 14.66 (Table 2), indicating preferential incorporation of lighter lanthanides relative to heaver lanthanides formed during partial melting⁵⁶. $\Sigma LREE/Th + U$ ranges from 5.70 to 28.33 (Table 1), which also supports substitutions of REE by Th and U (ref. 46).

The regression correlation analysis was carried out for chemical elements (Table 3) of monazites to understand the relation among different elements. LREE (La and Ce) showed significant positive correlation with calcium. Phosphorus has significant negative correlation with thorium and silicon. The HREEs such as Gd, Tb, Dy, Ho and Er showed significant positive correlation with yttrium.

Coupled ionic substitution such as $Ca^{+2} + Th^+ \Rightarrow 2REE^{+3}$ and $Th^{+4} + Si^{+4} \Rightarrow P^{+5} + REE^{+3}$ was common. In order to understand these relationships, regression analysis was carried out between Th + Ca with LREE, REE and REE + Y and Th + Si with LREE + P, REE + P and REE + P + Y (Table 4) and which shows inverse relationships.

The correlation coefficients between (Th + Si) versus (P + LREE), (P + REE) and (P + REE + Y) of coastal sands showed negative relation amongst the X_i and Y_i components, which indicates substitution of these elements at major cation structural sites. The correlation coefficients of Th + Ca versus $\Sigma LREE$, ΣREE and $\Sigma REE + Y$ of costal sands showed significant negative correlation and inverse relationship amongst X_i and Y_i components, indicating substitution (Table 4). These correlation coefficients show that Th + Si behaved similar to Th + Ca with similar regression coefficients. Thus, it indicates that Th + Si substitute P + LREE is another prime substitution in the analysed samples. The ratio $(La/Sm)_{Cn}$ in monazites varies from 1.60 to 10.33, which indicates that extended fractionation among the lighter lanthanides is less.

In the monazites from the coastal sands of Bhimunipatnam–Konada, Nd/Ce ratio varies from 0.34 to 0.39 and La/Ce varies from 0.38 to 0.57 (Table 2). These values are comparable to the values of monazites from Kalingapatnam–Baruva coast, Andhra Pradesh²⁴ and those from khondalite⁴⁵ and charnockites^{43,46} of the Eastern Ghats of India as well as monazites from Eastern Minasas Gerias, Brazil⁵⁷.

Ce > Nd > La > Sm > Pr > Gd pattern is common in monazites⁵⁸. The distribution pattern of REE in monazites

							Ε	able 1.	Chem	cal dat	ı of moı	nazites	from B	himuni	oatnam	to Kona	da coas	tal sanc	ls (wt%	~							
Sample no.	CS1	CS2	CS	CS4	CS5	CS6	CS7	CS8	CS9	CS10	CS11	CS12	CS13 C	S14 C	S15 C	S16 C	S17 CS	118 CS	19 CS2	0 RS2	1 RS2	2 RS23	RS24	R S25	Min.	Max.	Av.
SiO	0 79	0.85	1 63	2.85	1 07	1 10	0 66	1 00	1 04	0.58	0 50	0.53	0 57 (0 57 0	0 65 0	59 0	50 0	53 0	57 0.5	2 0 2	3 04	8 0.51	1 54	1 19	0 48	2.85	0.85
CaO	1 46	0.75	1 24	1 66	1.31	1 34	1 75	1 2.9	1.21	1 29	1 52	1 47	0 71	1 76 0	81 0	81	52 1	47 0	1 12	9 1 5	2 1 3	9 1 38	1 2.6	0.81	0 71	1 76	1 29
no,	0.28	0.08	0.05	0.03	0.08	0.07	0.15	0.09	0.10	0.42	0.09	0.09	0.07	0.15 0	08 0	08 0	0 60	60	07 0.1	5 0.3	1 0.2	3 0.23	0.06	0.04	0.03	0.42	0.13
ThO,	9.02	8.36	12.00	13.39	10.05	10.12	11.86	11.16	11.85	11.00	10.88	10.12	8.87 10	01 8	12	78 8	57 8	59 3	98 10 (1 8 6	7.5	8 7.56	11.75	8.32	3.78	3 39	9.42
PbO	0.42	0.24	0.53	0.59	0.44	0.43	0.42	0.25	0.25	0.36	0.39	0.39	0.18 (0.41 0	.17 0	17 0	39 0.	39 0	18 0.4	1 0.4	2 0.3	7 0.37	0.51	0.37	0.17	0.59	0.36
P_2O_5	30.20	28.89	27.94	28.88	29.89	29.97	29.40	28.67	28.55	28.14	30.12	29.55	9.19 30	0.79 29	20 29	20 29	33 29.	55 29	19 29.4	1 30.5	3 30.0	5 30.38	28.41	28.4	27.94 3	60.79	29.36
$Y_{2}O_{3}$	2.26	0.05	0.57	0.12	0.11	0.12	0.07	0.03	0.03	2.21	0.05	0.05	0.00 (0.06 0	0.01 0	.01 0	05 0.	05 0.	0.0 0.0	6 2.2	0.0	9 0.09	0.61	0.03	0.00	2.26	0.36
La_2O_3	10.50	17.18	11.55	8.79	11.39	10.40	10.50	13.25	13.12	12.15	11.19	11.21	2.45 10	0.38 13	.47 13	.47 10	60 11.	21 12	45 10.3	8 9.7	3 12.5	3 12.4	11.87	13.72	8.79	7.18	11.83
Ce_2O_3	25.58	29.93	25.32	23.22	26.06	26.13	25.51	26.38	26.09	26.17	26.17	27.11	9.09 2	7.77 28	.96 28	.96 26	17 27.	11 29	09 25.2	3 23.7	5 28.2	1 27.97	25.25	28.3	23.22	9.93	26.78
Pr_2O_3	2.85	2.50	2.61	2.94	2.96	2.93	2.98	2.54	2.56	3.21	3.58	2.96	3.22	2.89 2	.94 2	.94 2	92 2.	96 3.	22 2.8	9 2.7	2 3.0	9 2.93	2.63	2.8	2.50	3.58	2.91
Nd_2O_3	10.51	7.90	9.73	11.40	11.13	10.94	10.80	9.99	10.15	11.03	10.42	10.69	1.59 1(0.73 11	.03 11	.03 10.	42 10.	69 11	59 10.7	3 9.6	9 10.7	7 10.71	9.77	9.89	7.90	1.59	10.53
Sm_2O_3	1.74	0.49	1.33	1.57	1.31	1.31	1.32	1.05	0.98	1.14	1.22	1.15	1.30	1.36 1	.44	44	22 1.	15 1	30 1.3	6 1.8	1.1	1 1.08	1.36	0.78	0.49	1.80	1.25
Eu_2O_3	0.53	0.51	0.54	0.66	0.59	0.63	0.62	0.56	0.57	0.56	0.62	0.59	0.75 (0.61 0	.69 0	.0 69.	62 0.	59 0	.75 0.6	0.5	9 0.6	0.59	0.54	0.57	0.51	0.75	0.61
Gd ₂ O ₃	1.20	0.20	0.62	0.45	0.45	0.42	0.23	0.18	0.19	0.23	0.18	0.20	0.16	0.29 0	.15 0	.15 0	18 0.	20	.16 0.2	1.1	9 0.2	0.38	0.82	0	0.00	1.20	0.35
Tb_2O_3	0.21	0.01	0.06	0.09	0.00	0.00	0.00	0.02	0.00	0.02	0.00	0.02	0.00	0.02 0	01 0	010	00 0.	02	00 0.0	0.2	0.0	0	0.11	0	0.00	0.21	0.03
Dy_2O_3	0.67	0.05	0.20	0.12	0.07	0.12	0.05	0.03	0.04	0.06	0.05	0.00	0.00	0.05 0	00.00	00.	05 0.	00	00 0.0	15 0.6	5 0.0	7 0.05	0.19	0.05	0.00	0.67	0.10
Ho_2O_3	0.49	0.05	0.23	0.16	0.14	0.13	0.10	0.08	0.06	0.08	0.12	0.10	0.03 (0 60.0	.11 0	.11 0	12 0.	10 0.	.03 0.0	9 0.3	9 0.1	3 0.15	0.24	0.06	0.03	0.49	0.13
Er_2O_3	0.12	0.00	0.01	0.01	0.02	0.00	0.01	0.00	0.00	0.01	0.03	0.00	0.01 (0.02 0	.04 0	.04 0	03 0.	00	.01 0.0	0.1	5 0	0.03	0.07	0	0.00	0.15	0.03
Tm_2O_3	0.18	0.04	0.13	0.09	0.12	0.14	0.10	0.10	0.06	0.12	0.10	0.09	0.12 (0.15 0	.12 0	.12 0	10 0.	0 60	.12 0.1	5 0.1	6 0.0	7 0.12	0.15	0.08	0.04	0.18	0.11
Yb_2O_3	0.03	0.00	0.00	0.00	0.00	0.03	0.01	0.00	0.03	0.03	0.00	0.00	0.01 (0.01 0	.04 0	.04	00 0.	0 00	.01 0.0	1 0.0	1 0.0	1 0	0.04	0.01	0.00	0.04	0.01
Lu_2O_3	0.00	0.02	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00 0	0 00.	0 00	00 0.	00	00 0.0	0 0	0.0	1 0	0	0	0.00	0.02	0.00
Total	99.01	98.08	96.26	97.01	97.21	96.32	96.51	99.96	96.96	98.80	97.22	96.32	8.30.98	8.09 97	.96 93	.64 92	88 94.	79 93	43 94.2	6 95.1	4 97.0	6 96.93	97.18	95.42	92.88	9.01	96.46
Number o	f ration	and the	, hacie	of A over	vran atr	our																					
Si	1 Laure	15 ULU ULU 0 03		יי 11 0 11	у <u>вси а</u> и 0.04	61110 0 04	0.03	0.04	0.04	0.02	0.02	0.02	0.02	0.02	0.2	02_0	0.2 0	0.20	0.2 0.0	0 0 0	0 0 0	2 0 02	0.06	0.05	0.02	0 11	0.03
5 5	0.06	0.03	0.05	0.07	0.06	0.06	0.08	0.06	0.06	0.06	0.06	0.06	0.03 (0 20.0	0 20	104	0 20	0 90	03 00	200	0.0 10 10	4 0.06 A	0.05	0.04	20.0	0.08	0.06
	00.00	0.00	00.0	0.00	00.00	00.00	00.00	0.00	00.00	00.00	00.00	0.00	00.0	00.00	00.00	00	00		00 00	000	0,00	0.00	0.0	0.01	00.0	00.00	00.00
Th	0.08	0.08	0.11	0.12	0.09	0.09	0.11	0.10	0.11	0.10	0.10	0.09	0.08 (0 60.0	0.07 0	.03 0	08 0.	08 0	0.0 0.0	0.0 6	8 0.0	7 0.07	0.11	0.08	0.03	0.12	0.09
Pb	0.00	0.00	0.01	0.01	00.00	0.00	0.00	0.00	0.00	00.00	0.00	0.00	0.00 (0.00 0	0 00.	00.00	00 0.	00 00	0.0 0.0	0 0	0	0	0.01	0	00.00	0.01	0.00
Ь	0.99	0.98	0.96	0.96	1.00	1.01	1.00	0.98	0.98	0.96	1.01	1.01	0.99	1.02 0	1 66.	.00	01 1.	00 1	01 1.0	0 1.0	1	-	0.96	0.97	0.96	1.02	0.99
Y	0.05	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.05	0.00	0.00	0.00 (0.00 0	0 00.0	0 00	00 00	0 00	0.0 0.0	0.0 0.0	5 0	0	0.01	0	0.00	0.05	0.01
La	0.15	0.25	0.17	0.13	0.17	0.15	0.16	0.20	0.20	0.18	0.16	0.17	0.18 (0.15 0	.20 0	20 0	16 0.	17 0	19 0.1	5 0.1	4 0.1	8 0.18	0.17	0.21	0.13	0.25	0.17
Ce	0.36	0.44	0.38	0.33	0.38	0.38	0.38	0.39	0.39	0.39	0.38	0.40	0.43 (0.40 0	.42 0	.43 0	39 0.	40 0.	43 0.3	7 0.3	4 0.4	1 0.4	0.37	0.42	0.33	0.44	0.39
Pr	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.05	0.05	0.04	0.05 (0.04 0	.04 0	.04	04 0.	04 0	.05 0.0	14 0.0	4 0.0	4 0.04	0.04	0.04	0.04	0.05	0.04
PQ 5	0.15	0.11	0.14	0.16	0.16	0.15	0.16	0.14	0.15	0.16	0.15	0.15	0.17 ().15 0 	0.16	.16 20	.15 0. 22 0.	15 0. 0.	.17 0.1	5 0.1	4 0.1 0.1	5 0.15	0.14	0.14	0.11	0.17	0.15
Sm 1	20.0	10.0	70.0 10.0	10.0	0.02	20.0	0.02	10.0	10.0	70.0	20.0	20.0	20.0	0.020.0	0 70.	70.0	07 0.	0 70	0.0 10 10 20.			10.0 2	10.0	0.01	10.0	10.0	70.0
n d	10.0	0.00	10.0	10.0	10.0	10.0	10.0	0.00	00.0	0.00	0.00	10.0										10.0	10.0	10.0	10.0	10.0	10.0
n f	70°0	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00.0											0.0		0.00	70.0	0.00
Dv Dv	0.01	0.00	0.00	0.00	00.0	00.00	00.00	0.00	0.00	0.00	0.00	0.00	00.00	0000	000		00 00		00 00		0 0			0 0	00.0	0.01	0.00
Ho	0.01	0.00	00.0	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00.00	00	00 0.	00	0.0 0.0	0	0	0	0	0	0.00	0.01	0.00
Er	0.00	0.00	0.00	0.00	00.00	00.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00 (0.00 0	00.00	00	00 0.	00	0.0 0.0	0 0	0	0	0	0	00.00	0.00	0.00
Tm	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00 (0.00 0	00.0	0 00.	00 0.	0 00	0.0 0.0	0 0	0	0	0	0	0.00	0.00	0.00
Yb	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00.00	0.00	0.00	0.00	0.00	00.0	00.00	00 0.	0 00	00 0.0	0 0	0	0	0	0	0.00	0.00	00.00
Lu	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00 0	00.00	0 00.	00 0.	00 00	00 00	0 0	0	0	0	0	0.00	0.00	0.00
Total	1.99	1.99	1.95	1.97	1.98	1.97	1.98	1.98	1.98	2.00	1.98	1.98	1.98	1.98 1	.99 1	.95 1	95 1.	95 1	97 1.9	3 1.9	4 1.9	6 1.94	1.96	1.97	1.93	2.00	1.97
$\sum \text{LREE}$	8.81	11.03	6.72	5.70	8.35	8.12	6.81	7.58	7.10	7.53	7.67	8.34	0.32	8.37 11	.30 28	.33 9.	50 9.	75 21	50 8.1	1 8.5	0 11.4	3 11.1	6.72	10.25	5.70	8.33	96.6
ΣLREE/Υ	, 15	780	60	266	320	289	537	1458	1342	16	782	686 9	864 6	26 43	IO 66	V D	IV DIV	DI DI	V DIV	/ 13.	DIV	DIV	74	DIV	13	9864.	1266
CS Coast	al cando	•. RS R	A Ser	1 ments		i			1	(1.2	2						I.			ł			1	ł		
1000, CU	dl Salıu	N, NO, N	Con sol	IIIICIIIS.																							

Table 2. Chemical data of monazites from Bhimunipatnam to Konada coastal sands

Sample no.	ΣLREE	Th + U (Acti)	Σree	ZHREE	SLREE/ HREE	REE + P	REE + P + Y	LREE + P	Nd/Ce	La/Ce	Th + Si	Th/U	Th + Ca	REE + Y	(La/Sm)cn	(ZLLn/HLn)cn
CS1	51.18	9.29	54.6	3.42	14.97	84.8	87.05	81.38	0.41	0.41	9.6	32.31	10.48	56.85	1.79	4.13
CS2	58	8.44	58.87	0.87	67.05	87.76	87.81	86.89	0.26	0.57	9.21	101.96	9.11	58.92	10.33	14.66
CS3	50.53	12.04	52.31	1.78	28.39	80.25	80.82	78.47	0.38	0.46	13.63	255.26	13.24	52.88	2.59	6.86
CS4	47.92	13.42	49.48	1.57	30.6	78.36	78.48	76.8	0.49	0.38	16.25	432.03	15.05	49.61	1.66	7.17
CS5	52.85	10.13	54.26	1.41	37.62	84.15	84.26	82.74	0.43	0.44	11.12	124.07	11.36	54.37	2.57	8.47
CS6	51.7	10.19	53.16	1.46	35.38	83.13	83.25	81.67	0.42	0.4	11.22	136.74	11.46	53.28	2.35	7.87
CS7	51.1	12.01	52.21	1.11	46.21	81.61	81.67	80.5	0.42	0.41	12.51	77.51	13.61	52.27	2.36	9.92
CS8	53.21	11.25	54.17	0.96	55.6	82.83	82.86	81.88	0.38	0.5	12.16	119.96	12.45	54.19	3.75	10.85
CS9	52.89	11.94	53.85	0.96	55.32	82.4	82.42	81.44	0.39	0.5	12.88	122.11	13.15	53.87	3.99	12.4
CS10	53.7	11.42	54.81	1.11	48.46	82.95	85.16	81.84	0.42	0.46	11.57	26	12.29	57.02	3.16	9.86
CS11	52.57	10.97	53.67	1.1	47.79	83.79	83.84	82.69	0.4	0.43	11.38	122.24	12.4	53.72	2.73	10.14
CS12	53.12	10.2	54.12	1	53.02	83.67	83.72	82.67	0.39	0.41	10.65	119.05	11.59	54.18	2.9	10.37
CS13	57.64	8.94	58.71	1.07	53.71	87.9	87.9	86.83	0.4	0.43	9.44	126.7	9.58	58.71	2.85	10.39
CS14	53.13	10.16	54.35	1.22	43.59	85.14	85.19	83.92	0.39	0.37	10.58	68.58	11.78	54.41	2.27	8.58
CS15	57.83	8.2	58.98	1.16	50.07	88.19	88.19	87.03	0.38	0.47	8.71	106.83	8.93	58.99	2.78	9.75
CS16	57.84	3.86	59	1.16	49.86	88.2	88.21	87.04	0.38	0.47	4.37	47.25	4.59	59.01	2.78	9.68
CS17	51.33	8.66	52.43	1.1	46.66	81.76	81.81	80.66	0.4	0.41	9.07	95.22	10.09	52.48	2.58	9.53
CS18	53.12	89.8	54.12	1	53.12	83.67	83.72	82.67	0.39	0.41	9.12	95.44	10.06	54.17	2.89	10.39
CS19	57.65	4.05	58.73	1.08	53.38	87.92	87.92	86.84	0.4	0.43	4.55	56.86	4.69	58.73	2.84	10.27
CS20	50.59	10.16	51.83	1.24	40.8	81.3	81.36	80.06	0.43	0.41	10.58	66.73	11.77	51.89	2.27	8.24
RS21	47.69	8.91	51.03	3.34	14.28	81.56	83.76	78.22	0.41	0.41	9.13	27.74	10.12	53.23	1.6	4.13
RS22	55.71	7.81	56.87	1.16	48.03	86.92	87.01	85.76	0.38	0.44	8.06	32.96	8.97	56.96	3.35	69.6
RS23	55.09	7.79	56.41	1.32	41.73	86.79	86.88	85.47	0.38	0.44	8.07	32.87	8.94	56.5	3.41	8.99
RS24	50.88	11.81	53.04	2.16	23.56	81.45	82.06	79.29	0.39	0.47	13.29	195.83	13.01	53.65	2.59	5.87
RS25	55.49	8.36	56.26	0.77	72.06	84.66	84.69	83.89	0.35	0.48	9.51	208	9.13	56.29	5.2	13.1
Min.	42.9	3.81	43.47	0.57	75.26	71.41	71.41	70.84	0.34	0.38	4.26	121.94	4.49	43.47	1.6	4.13
Max.	64.08	13.82	67.78	3.7	17.32	98.56	100.82	94.86	0.39	0.57	16.25	31.66	15.16	70.03	10.33	14.66
Av.	53.31	9.55	54.69	1.38	38.63	84.05	84.4	82.67	0.39	0.44	10.27	74.16	10.71	55.05	3.1	9.25
CS, Costal	sands; RS,	Red sedim	ents.													

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Lu	1.00
Чþ	1.00 0.05
Tm	1.00 0.54 0.13
Er	1.00 0.59 0.51 -0.02
Но	1.00 0.85 0.77 0.03
Dy	$\begin{array}{c} 1.00\\ 0.51\\ 0.51\\ 0.21\\ 0.19\\ -0.06\end{array}$
$\frac{(n=25)}{\mathrm{Tb}}$	1.00 0.57 0.89 0.81 0.51 0.31 0.31
Konada Gd	1.00 0.51 0.90 0.44 0.28 0.16
atnam to Eu	$\begin{array}{c} 1.00\\ -0.27\\ -0.49\\ -0.43\\ -0.71\\ -0.35\\ -0.47\end{array}$
shimunip Sm	1.00 0.23 0.34 0.35 0.31 0.35 0.34 0.35 0.34 0.35 0.34 0.15
ds from E Nd	1.00 0.53 0.53 0.17 0.17 0.17
oastal san Pr	$\begin{array}{c} 1.00\\ 0.62\\ 0.62\\ 0.12\\ -0.11\\ -0.14\\ 0.03\\ 0.11\\ 0.24\\ 0.09\\ 0.03\end{array}$
azites of c Ce	$\begin{array}{c} 1.00\\ 0.08\\ 0.08\\ 0.18\\ 0.18\\ 0.18\\ 0.18\\ 0.18\\ 0.18\\ 0.21\\ 0.21\\ 0.17\\ 0.17\end{array}$
rix for mon La	$\begin{array}{c} 1.00\\ 0.81\\ -0.20\\ -0.39\\ -0.10\\ -0.33\\ -0.39\\ -0.39\\ -0.24\\ 0.02\\ 0.19\\ 0.02\\ 0.43\\ 0.02\end{array}$
ttion mat	$\begin{array}{c} 1.00\\ -0.28\\ -0.46\\ 0.04\\ 0.25\\ 0.49\\ 0.49\\ 0.42\\ 0.42\\ 0.42\\ 0.25\\ 0.25\\ 0.25\\ 0.25\\ 0.25\\ 0.23\\ 0.25\\ 0.2$
Correls P	$\begin{array}{c} 1.00\\ -0.16\\ 0.25\\ 0.25\\ 0.22\\ 0.22\\ 0.22\\ 0.22\\ 0.22\\ 0.22\\ 0.22\\ 0.22\\ 0.22\\ 0.07\\ 0.22\\ 0.07\\ 0.22$
lable 3. Pb	$\begin{array}{c} 1.00\\ -0.46\\ -0.46\\ -0.46\\ -0.07\\ -0.01\\ 0.15\\ 0.41\\ 0.41\\ 0.41\\ 0.47\\ 0.23\\ 0.55\\ 0.15\\ 0.15\\ 0.13\\ $
Th dT	$\begin{array}{c} 1.00\\ 0.69\\ 0.10\\ 0.12\\ 0.12\\ 0.17\\ 0.17\\ 0.12\\ 0.12\\ 0.12\\ 0.12\\ 0.12\\ 0.12\\ 0.12\\ 0.12\\ 0.12\\ 0.12\\ 0.03\\$
n	$\begin{array}{c} 1.00\\ 0.30\\ 0.34\\ 0.34\\ 0.34\\ 0.27\\ 0.08\\ 0.13\\ 0.27\\ 0.08\\ 0.19\\ 0.02\\ 0.19\\ 0.19\\ 0.19\\ 0.19\\ 0.20\\ 0.20\\ 0.40\\ 0.40\\ 0.67\\ 0.67\\ 0.67\\ 0.64\\ 0.61\\ 0.60\\ 0.61\\$
Ca	$\begin{array}{c} 1.00\\ 0.12\\ 0.50\\ 0.17\\ 0.24\\ 0.09\\ 0.01\\ 0.01\\ 0.01\\ 0.01\\ 0.01\\ 0.01\\ 0.01\\ 0.01\\ 0.01\\ 0.06\\ 0.11\\ 0.06\\ 0.01\\ 0.05\\ 0.02\\ 0.05\\ 0.02\\ 0.05\\ 0.02\\ 0.05\\$
Si	$\begin{array}{c} 1.00\\ 0.05\\ -0.10\\ 0.57\\ 0.59\\ 0.59\\ 0.59\\ 0.59\\ 0.59\\ 0.59\\ 0.59\\ 0.23\\ -0.10\\ 0.04\\ 0.04\\ 0.04\\ 0.04\\ 0.04\\ 0.04\\ 0.02\\ 0.02\\ 0.02\\ 0.02\\ 0.00\\ 0.0$
	Si Ca Th P P S M Ca S Ca S Ca S M Ca S M Ca S M Ca S Th Th Th Th Th Th Th Th Th Th Th Th Th

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Figure 2. Chondrite-normalized (chondrite values from Taylor and McLennan⁵⁴) patterns of REEs for monazite of coastal sands from Bhimunipatnam to Konada.

of the study area is Ce > La > Nd > Pr > Sm > Gd. Similar distribution patterns in monazite were observed in different litho units from the Eastern Ghats Granulite (EGG) Belt⁴⁶ and Kalingapatnam–Baruva coastal sands²⁴. However, the monazite grains contain greater LREE than HREE content and less yttrium concentration. In general, monazite associated with garnet has less affinity towards yttrium and HREE^{46,59}.

Figure 2 shows the chondrite-normalized REE distribution patterns of individual monazite grains from coastal sands and red sediments for the study area.

REE distribution pattern in monazites is influenced by geological evolution mainly mineral phase assemblage, and source-rock material from which the monazites were formed⁶⁰. The monazite sands of the study area showed positive europium (Eu) anomaly. The negative or positive

anomalies are a reflection of enrichment/depletion of Eu in progressive crystallization of magma or anatectic melt under high oxygen fugacity (more oxidation conditions) or reduction conditions⁵⁶. Magnetite is one of the important accessory minerals in the khondalite and charnockite suite of rocks, which indicates high oxidation conditions during crystallization of anatectic melt of khondalites and crystallization of magma for charnockites. In the more oxidation conditions of magma, europium is in the Eu^{+2} state, which does not incorporate in the plagioclase feldspars but at less oxidizing conditions, europium is in the Eu^{+2} state and it will enter into the plagioclase feldspars. So the enrichment of Eu⁺³ in the magma in relation to REE gives positive anomalies in the chondritenormalized REE patterns of monazites. In the Eastern Ghats Group of rocks, the magnetite content and REE patterns in charnockites of enderbites nature of Visakhapatnam region show Eu positive anomaly indicates that

 Table 4. Correlation coefficients of chemical data from Bhimunipatnam to Konada coastal sands

	Th + Ca		Th + Si
LREE (La–Sm)	-0.810	LREE + P $REE + P$ $REE + P + Y$	-0.763
REE	-0.827		-0.792
REE + Y	-0.824		-0.832

Table 5.	U-Th-Pb geochemical dating of monazites
from c	oastal sands of Bhimunipatnam–Konada

$\begin{array}{c ccccc} CS1 & 968 \pm 50 \\ CS2 & 835 \pm 57 \\ CS3 & 1003 \pm 47 \\ CS4 & 1014 \pm 46 \\ CS5 & 990 \pm 50 \\ CS6 & 984 \pm 50 \\ CS7 & 926 \pm 48 \\ CS8 & 1114 \pm 37 \\ CS9 & 1024 \pm 37 \\ CS10 & 1018 \pm 56 \\ CS11 & 1019 \pm 55 \\ CS12 & 1006 \pm 34 \\ CS13 & 1003 \pm 53 \\ CS14 & 904 \pm 87 \\ CS15 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS17 & 1019 \pm 56 \\ CS18 & 1007 \pm 54 \\ CS19 & 1003 \pm 83 \\ CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \\ \end{array}$	Sample no.	Age (Ma)
CS1F (0) \pm (5)CS2 835 ± 57 CS3 1003 ± 47 CS4 1014 ± 46 CS5 990 ± 50 CS6 984 ± 50 CS7 926 ± 48 CS8 1114 ± 37 CS9 1024 ± 37 CS10 1018 ± 56 CS11 1019 ± 55 CS12 1006 ± 34 CS13 1003 ± 53 CS14 904 ± 87 CS15 981 ± 84 CS16 981 ± 84 CS17 1019 ± 56 CS18 1007 ± 54 CS19 1003 ± 83 CS20 903 ± 47 RS21 1006 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34		968 + 50
CS2 003 ± 37 CS3 1003 ± 47 CS4 1014 ± 46 CS5 990 ± 50 CS6 984 ± 50 CS7 926 ± 48 CS8 1114 ± 37 CS9 1024 ± 37 CS10 1018 ± 56 CS11 1019 ± 55 CS12 1006 ± 34 CS13 1003 ± 53 CS14 904 ± 87 CS15 981 ± 84 CS17 1019 ± 56 CS18 1007 ± 54 CS19 1003 ± 83 CS20 903 ± 47 RS21 1006 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	CS2	835 + 57
CS31003 \pm 47CS41014 \pm 46CS5990 \pm 50CS6984 \pm 50CS7926 \pm 48CS81114 \pm 37CS91024 \pm 37CS101018 \pm 56CS111019 \pm 55CS121006 \pm 34CS131003 \pm 53CS14904 \pm 87CS15981 \pm 84CS16981 \pm 84CS171019 \pm 56CS181007 \pm 54CS191003 \pm 83CS20903 \pm 47RS211006 \pm 52RS221021 \pm 56RS231019 \pm 56RS24999 \pm 48RS251002 \pm 55Min.835 \pm 34	CS3	1003 ± 47
CS11014 \pm 40CS5990 \pm 50CS6984 \pm 50CS7926 \pm 48CS81114 \pm 37CS91024 \pm 37CS101018 \pm 56CS111019 \pm 55CS121006 \pm 34CS131003 \pm 53CS14904 \pm 87CS15981 \pm 84CS16981 \pm 84CS171019 \pm 56CS181007 \pm 54CS191003 \pm 83CS20903 \pm 47RS211006 \pm 52RS221021 \pm 56RS231019 \pm 56RS24999 \pm 48RS251002 \pm 55Min.835 \pm 34	CS4	1014 ± 46
CS5 936 ± 50 CS6 984 ± 50 CS7 926 ± 48 CS8 1114 ± 37 CS9 1024 ± 37 CS10 1018 ± 56 CS11 1019 ± 55 CS12 1006 ± 34 CS13 1003 ± 53 CS14 904 ± 87 CS15 981 ± 84 CS16 981 ± 84 CS17 1019 ± 56 CS18 1007 ± 54 CS19 1003 ± 83 CS20 903 ± 47 RS21 1006 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	CS5	990 ± 50
$\begin{array}{ccccc} & & & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & &$	CS6	984 ± 50
$\begin{array}{ccccc} S1 & 1126 \pm 437 \\ CS8 & 1114 \pm 37 \\ CS9 & 1024 \pm 37 \\ CS10 & 1018 \pm 56 \\ CS11 & 1019 \pm 55 \\ CS12 & 1006 \pm 34 \\ CS13 & 1003 \pm 53 \\ CS14 & 904 \pm 87 \\ CS15 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS17 & 1019 \pm 56 \\ CS18 & 1007 \pm 54 \\ CS19 & 1003 \pm 83 \\ CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \\ \end{array}$	CS7	926 ± 48
$\begin{array}{cccccc} 1114 \pm 37 \\ CS9 & 1024 \pm 37 \\ CS10 & 1018 \pm 56 \\ CS11 & 1019 \pm 55 \\ CS12 & 1006 \pm 34 \\ CS13 & 1003 \pm 53 \\ CS14 & 904 \pm 87 \\ CS15 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS17 & 1019 \pm 56 \\ CS18 & 1007 \pm 54 \\ CS19 & 1003 \pm 83 \\ CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \\ \end{array}$	CS8	1114 + 37
$\begin{array}{ccccc} 1024 \pm 5 \\ CS10 & 1018 \pm 5 \\ CS11 & 1019 \pm 5 \\ CS12 & 1006 \pm 34 \\ CS13 & 1003 \pm 5 \\ CS14 & 904 \pm 87 \\ CS15 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS17 & 1019 \pm 5 \\ CS18 & 1007 \pm 54 \\ CS19 & 1003 \pm 83 \\ CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \\ \end{array}$	CS9	1024 + 37
$\begin{array}{ccccc} \text{CS11} & 1019 \pm 55 \\ \text{CS12} & 1006 \pm 34 \\ \text{CS13} & 1003 \pm 53 \\ \text{CS14} & 904 \pm 87 \\ \text{CS15} & 981 \pm 84 \\ \text{CS16} & 981 \pm 84 \\ \text{CS17} & 1019 \pm 56 \\ \text{CS18} & 1007 \pm 54 \\ \text{CS19} & 1003 \pm 83 \\ \text{CS20} & 903 \pm 47 \\ \text{RS21} & 1006 \pm 52 \\ \text{RS22} & 1021 \pm 56 \\ \text{RS23} & 1019 \pm 56 \\ \text{RS24} & 999 \pm 48 \\ \text{RS25} & 1002 \pm 55 \\ \text{Min.} & 835 \pm 34 \\ \end{array}$	CS10	1024 ± 57 1018 ± 56
CS11 1019 ± 53 CS12 1006 ± 34 CS13 1003 ± 53 CS14 904 ± 87 CS15 981 ± 84 CS16 981 ± 84 CS17 1019 ± 56 CS18 1007 ± 54 CS19 1003 ± 83 CS20 903 ± 47 RS21 1006 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	CS11	1010 ± 50 1019 ± 55
CS12 1000 ± 53 CS13 1003 ± 53 CS14 904 ± 87 CS15 981 ± 84 CS16 981 ± 84 CS17 1019 ± 56 CS18 1007 ± 54 CS19 1003 ± 83 CS20 903 ± 47 RS21 1006 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	CS12	1006 ± 34
CS15 1000 ± 250 CS14 904 ± 87 CS15 981 ± 84 CS16 981 ± 84 CS17 1019 ± 56 CS18 1007 ± 54 CS19 1003 ± 83 CS20 903 ± 47 RS21 1006 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	CS13	1000 ± 54 1003 ± 53
CS14 304 ± 0 CS15 981 ± 84 CS16 981 ± 84 CS17 1019 ± 56 CS18 1007 ± 54 CS19 1003 ± 83 CS20 903 ± 47 RS21 1006 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	CS14	904 + 87
$\begin{array}{cccc} CS16 & 981 \pm 84 \\ CS16 & 981 \pm 84 \\ CS17 & 1019 \pm 56 \\ CS18 & 1007 \pm 54 \\ CS19 & 1003 \pm 83 \\ CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \\ \end{array}$	CS15	981 ± 84
$\begin{array}{cccc} CS17 & 1019 \pm 56 \\ CS18 & 1007 \pm 54 \\ CS19 & 1003 \pm 83 \\ CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \\ \end{array}$	CS16	981 ± 84
$\begin{array}{cccc} CS18 & 1007 \pm 54 \\ CS19 & 1003 \pm 83 \\ CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \\ \end{array}$	CS17	1019 ± 56
$\begin{array}{cccc} CS19 & 1003 \pm 83 \\ CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \\ \end{array}$	CS18	1007 ± 50
$\begin{array}{cccc} CS20 & 903 \pm 47 \\ RS21 & 1006 \pm 52 \\ RS22 & 1021 \pm 56 \\ RS23 & 1019 \pm 56 \\ RS24 & 999 \pm 48 \\ RS25 & 1002 \pm 55 \\ Min. & 835 \pm 34 \end{array}$	CS19	1003 ± 83
RS21 1006 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	CS20	903 ± 47
RS21 1000 ± 52 RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	RS21	1006 ± 52
RS22 1021 ± 56 RS23 1019 ± 56 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	RS22	1020 ± 52
RS25 1019 ± 30 RS24 999 ± 48 RS25 1002 ± 55 Min. 835 ± 34	RS23	1019 ± 56
RS25 1002 ± 55 Min. 835 ± 34	RS24	999 + 48
Min. 835 ± 34	RS25	1002 ± 55
	Min	835 + 34
Max 1114 + 87	Max	1114 + 87
Av. 990 ± 55	Av.	990 ± 55

CS, Coastal sands; RS, Red sediments.

charnockites protolith were formed under high oxygen fugacity of magma and most of charnockites of AP are calc-alkaline magmas, which are in general more oxidized³⁸. The monazite associated with charnockites and the monazite inclusions in garnets of the khondalites show positive Eu anomaly, but the monazites in pyroxene granulites show negative Eu anomalies⁴⁵. The positive Eu anomaly of monazites in the study area indicates that these are derived from charnockites and tiny inclusions in garnet occurring in the khondalites.

Table 5 provides data on the U–Th–Pb geochemical dating of monazite grains of coastal sands carried out in the present study. The age of monazite grains of coastal sands ranges from 835 ± 34 to 1114 ± 87 Ma.

The age of monazites from various rock types of the EGG Belt of India ranges from 1600 to 500 Ma, in response to tectonic–metamorphic events. Average age of studied monazite grains is 1000 Ma. Similar ages for monazites were noted in khondalites (990–965 Ma) and charnockite complex (980–910 Ma) of the respect drainage basins of the Gosthani and Champavathi rivers⁴² and adjacent drainage basins of the Nagavali River in the study area, which is part of zone-II and zone-III isotopic domains of the Eastern Ghats⁶¹.

In the Eastern Ghats two ages for protoliths emplacement have been recorded⁶², i.e. middle Proterozoic ~1000 Ma. The charnockites of the Eastern Ghats are subjected to two periods of protolithic intrusions, one in palaeo-Proterozoic and the other in meso-neo Proterozoic⁶². Therefore, the monazite grains might be derived from charnockites formed during meso-neo Proterozoic and/or garnet-bearing metapelitic rock (khondalites) which are formed in high-grade metamorphism during the same time.

Thus ThO₂ content in monazites of Bhimunipatnam– Konada coastal sands varies from 3.78% to 13.39% with an average of 9.42%, indicating these are monazites and not cheralites. ThO₂ content indicates that almost 50% of the studied grains are from metapelitic rocks and the remaining are from the charnockite suite of rocks.

All the monazites studied show characteristic feature that the total REE is more than actinides (U + Th). In general, the garnet-bearing paragenesis rocks the HREE preferred to enter into garnets along with yttrium than other minerals.

The greater LREE content compared to HREE content with less yttrium concentration in monazites of coastal sands indicates that it is derived from metapelitic khondalites and charnokites in the study area, showing garnetbearing paragenesis.

The U–Th–Pb geochemical dating of monazites in the study area indicates that the age of monazite grains is about 1000 Ma. These grains might be derived from second phase (meso–neo-Proterozoic) of protolithic intrusions, i.e. charnockites which are present surrounding the study area and metapelitic rocks formed in the period

950–1100 Ma of the Main Eastern Ghats Orogeny (Grenville) accompanied by pervasive metamorphism, which causes monazite formation in khondalites of the same age.

- Mohanty, A. K., Das, S. K., Vijayan, V., Sengupta, D. and Saha, S. K., Geochemical studies of monazite sands of Chatrapur beach placer deposit of Orissa, India by PIXE and EDXRF method. *Nucl. Instr. Methods B*, 2003, 211, 145–154.
- Panda, N. K., Rajagopalan, V. and Ravi, G. S., Rare earth-element geochemistry of placer monazites from Kalingapatnam Coast, Srikakulam district, Andhra Pradesh. J. Geol. Soc. India, 2003, 62(4), 429–438.
- Mahadevan, C. and Sriramadas, A. S., Monazites in the beach sands of Vizagapatam District. *Proc. Indian Acad. Sci. Sect. A*, 1948, 27, 275–278.
- 4. Mahadevan, C. and Sathapathi, N., The home of monazite in Vizagapatam area. *Curr. Sci.*, 1948, **17**, 297.
- 5. Mahadevan, C. and Rao, N. B., Black sand concentrates of the Vizagapatam coast. *Curr. Sci.*, 1950, **19**, 48–49.
- Sriramadas, A., The black sand concentrates of the Vizagapatam Beach. Q. J. Geol. Min. Metall. Soc. India, 1951, 23,169–180.
- 7. Borreswar Rao, C., Beach erosion and concentration of heavy minerals. J. Sediment. Petrol., 1957, 27, 143–147.
- Mahadevan, C., Narayan Das, G. R. and Nagaraja Rao, N., Prospecting and evaluation of beach placers along the coastal belt of India. In Proceedings of the 2nd International Conference on Peaceful Uses of Atomic Energy, Geneva, 1958, vol. 2, pp. 103– 106.
- Krishna Rao, J. S. R., Ore microscopic investigation of black sands concentrations of Visakhapatnam. *Sci. Cult.*, 1964, **30**, 552– 553.
- Swamy, A. S. R. and Srihari, Y., Radioactivity of beach sands and coastal sands and coastal dunes of the Bheemunipatnam area, Andhra Pradesh. In Second convention. Indian Association of Sedimentologists, 1979, p. 15.
- Rao, A. T. and Prakasa Rao, Ch. S., Cheralite from Beach placers of Visakhapatnam–Bhimunipatnam coast. *Indian J. Mar. Sci.*, 1980, 9, 214–216.
- Sastry, A. V. R., Swamy, A. S. R. and Prasada Rao, Distribution of sands along Visakhapatnam–Bhimunipatnam beach. *Indian J. Mar. Sci.*, 1981, 1, 369–370.
- Ramamohana Rao, T., Shanmukha Rao, Ch. and Sanyasi Rao, K., Textural analysis and mineralogy of the black sand deposits of Visakhapatnam–Bhimunipatnam coast, AP, India. J. Geol. Soc. India, 1982, 23, 284–289.
- Jagannadha Rao, M. and Krishna Rao, J. S. R., Textural and mineralogical studies on red sediments of Visakhapatnam– Bhimunipatnam coast. *Geo. Views*, 1984, 2, 57–64.
- Swamy, A. S. R. and Krishna Rao, G., Radioactive intensities of beach sands of Bheemunipatnam area, Andhra Pradesh. *Geol. Soc. India Mem.*, 1985, 5, 139–144.
- Sastry, A. V. R., Swamy, A. S. R. and Vasudev, K., Heavy minerals of beach sands along Visakhapatnam–Bhimunipatnam, east coast of India. *Indian J. Mar. Sci.*, 1987, 16, 39–42.
- Ravi, G. S., Gajapathi Rao, R. and Yugandhara Rao, A., Coastal heavy mineral sand deposits of Andhra Pradesh. Special Issue on Beach and Inland Heavy Mineral Sand Deposits of India. *Atom. Min. Director. Explor. Res.*, 2001, 13, 53–85.
- Jagannadha Rao, M., Venkata Ramana, J., Venugopal, R. and Chandra Rao, M., Geochemistry and ore mineralogy of ilmenite from beach placers of the Visakhapatnam–Bhimunipatnam deposit, Andhra Pradesh. J. Geol. Soc. India, 2005, 66, 147–150.
- 19. Cheepurupalli, Rao, N., Anu Radha, B., Reddy, K. S. N., Dhanamjaya Rao, E. N. and Dayal, A. M., Heavy mineral distribution

studies in different microenvironments of Bhimunipatnam coast, Andhra Pradesh, India. *Int. J. Sci. Res. Publ.*, 2012, **2**, 1–10.

- Krishnaiah Setty, B. and Dhana Raju, R., Magnetite content as a basis to estimate other major heavy minerals content in the sand deposit along Nizampatnam coast, Guntur district, Andhra Pradesh. J. Geol. Soc. India, 1988, **31**, 491–494.
- Kshira Sagar, T. V. S. R. and Nagamalleswara Rao, B., Radiometric studies for monazite along the coastal region of west Godavari, east coast of India. *Indian J. Mar. Sci.*, 1989, 18, 8–10.
- 22. Kshira Sagar, T. V. S. R. and Nagamalleswara Rao, B., Magnetic susceptibility of black sand concentrate of Krishna district, east coast of India. *Indian J. Mar. Sci.*, 1991, **20**, 292–293.
- 23. Nagamalleswara Rao, B., Radiometric, magnetic susceptibility and mineralogical studies of some beach placers of Andhra Pradesh east coast of India. *J. Geol. Soc. India*, 1994, **43**, 669–675.
- Rajasekhara Reddy, D. and Siva Sankar Prasad, V., Thorium-rich monazites from the beach sands of Kalingapatnam–Baruva coast, Andhra Pradesh, east coast of India. *Curr. Sci.*, 1997, 73(10), 880– 882.
- Rajasekhara Reddy, D., Malathi, V., Reddy, K. S. N. and. Varma, D. D., Heavy minerals in different environments of beaches between Pudimadaka and Pentakota, east coast of India. *Indian Acad. Sci.*, 1998, 41(2), 47–54.
- Reddy, K. S. N., Lakshmi Prasad, T. and Babu Rao, N., Relationship of heavy mineral redistribution in different microenvironments to seasonal changes of beach processes in an embayed beach of Yarada–Gangavaram, north coastal Andhra Pradesh. *J. Geol. Soc. India*, 2007, **70**, 963–974.
- Reddy, K. S. N., Varma, D. D., Dhanamjaya Rao, E. N., Veeranarayana, B. and Lakshmi Prasad, T., Distribution of heavy minerals in Nizampatnam–Lankavanidibba coastal sands, Andhra Pradesh, east coast of India. J. Geol. Soc. India, 2012, 79, 411–418.
- Dhanunjaya Rao, D., Krishnaiah Setty, B. and Rami Naidu, Ch., Heavy mineral content and textural characteristics of coastal sands in the Krishna–Godavari, Gosthani–Champavathi, and Penna river deltas of Andhra Pradesh, India: a comparative study. *Explor. Res. At. Miner.*, 1989, 2, 147–155.
- 29. Yugandhar Rao, A., Ravi, G. S., Gajapathi Rao, R., Krishnan, S., Ali, M. A. and Banerjee, D. C., Srikurmam – a major heavy mineral beach and dune sand deposit on the east coast of India. In *Handbook of Placer Mineral Deposits* (ed. Rajamanickam, G. V.), New Academic Publishers, New Delhi, 2001, pp. 53–63.
- Behera, P., Heavy minerals in beach sands of Gopalpur and Paradeep along Orissa coast line, east coast of India. *Indian J. Mar. Sci.*, 2003, 32, 172–174.
- Sulekha Rao, N. and Misra, S., Source of monazite sand in southern Orissa beach placer, eastern India. J. Geol. Soc. India, 2009, 74, 357–362.
- Acharya, B. C., Nayak, B. K. and Das, S. K., Heavy mineral placer sand deposits of Kontiagarh area, Ganjam district, Orissa, India. *Resour. Geol.*, 2008, **59**(4), 388–399.
- Sanuprava, M., Behra, P. and Das, S. K., Heavy mineral potentiality and alteration studies for ilmenite in Astaranga beach sands, district Puri, Odisha, India. J. Geosci. Environ. Prot., 2015, 3, 31– 37.
- Babruvahana Rao, K., Origin and evolution of sand dune deposits of Ganjam coast, Orissa, India. *Explor. Res. At. Miner.*, 1989, 2, 133–146.
- 35. Rao, R. G., Sahoo, P. and Panda, N. K., Heavy mineral sand deposit of Orissa. *Explor. Res. At. Miner.*, 2001, **13**, 23–52.
- Divakara Rao, V., Khondalites from the Eastern Ghats granulite belt geochemistry and origin. *Geophys. Res. Bull.*, 1984, 21(3), 233-242.
- 37. Divakara Rao, V. and Murty, N. N., Evolution of the Eastern Ghats Granulite Belt, India. Precambrian crustal processes in Eastern Ghats granulite–greenstone regions of India and Antarctica within East Gondwana. In Gondwana Research Group, Mem.

4 (eds Rao, A. T., Divi, S. R. and Yoshida, M.), Field Sciences Publisher, 1998, pp. 1–18.

- Mallikharjuna Rao, M., Divakara Rao, V. and Murthy, N. N., Charnockite from EGMB – geochemistry, protoliths character and tectonic relevance. In Gondwana Research Group Mem. 5, 1999, pp. 31–50.
- Murthy, N. N. and Divakara Rao, V., Geochemistry, provenance and depositional environment of the khondalites from the Eastern Ghats Mobile Belt. In *Eastern Ghats Granulites* (eds Rao, A. T., Divakara Rao, V. and Yoshida, M.), Mem. 5, Field Sciences Publisher, 1999, pp. 15–30.
- Murthy, M. S., Occurrence of monazite in the charnockite of Visakhapatnam. *Curr. Sci.*, 1958, 27, 347–348.
- 41. Narayana, B. L., Rama Rao, P., Reddy, G. L. N. and Divakara Rao, V., Geochemistry and origin of megacrystic charnockites and granites from Eastern Ghat Granulite Belt. In Proceedings of the Symposium on India and Antarctica during the Precambrian and Granulite and Crustal Processes in East Gondwana, Andhra University, Visakhaptanam, 1–3 December 1995, p. 36.
- Simmat, R. and Raith, M. M., U–Th–Pb monazite geochronometry of the Eastern Ghats belt, India: timing and spatial disposition of poly-metamorphism. *Precambrian Res.*, 2008, 162, 16–39.
- Rao, A. T., Fonarev, V. I., Konilov, A. N. and Romanenko, M., Electron microprobe dating monazite from Spinel Granulite in the Eastern Ghats belt, India. J. Geol. Soc. India, 1998, 52, 345–350.
- 44. Rao, A. T., Rao, J. U. and Yoshida, M., Geochemistry and tectonic evolution of the pyroxene granulites from Visakhapatnam area in the Eastern Ghats Granulite Belt, India. J. Geosci., Osaka City Univ., 1993, 36, 135–150.
- Kamineni, D. C. and Rao, A. T., Sapphirine-granulites from the Kakanuru area, Eastern Ghats, India. *Am. Mineral.*, 1988, **73**, 692– 700.
- Kamineni, D. C., Rao, A. T. and Bonardi, M., The geochemistry of monazite types from the Eastern Ghats Granulite terrain, India. *Mineral. Petrol.*, 1991, 45, 119–130.
- 47. Sreenivasa Rao, P., Satyanarayana, G. and Swamy, A. S. R., Heavy minerals of modern and relict sediments of the Nizampatnam bay, East Coast of India. *Indian J. Mar. Sci.*, 1995, 24, 166– 170.
- Ali, M. A., Krishnan, S. and Banerjee, D. C., Beach and inland heavy mineral sand investigations and deposits in India – an overview. *Explor. Res. At. Miner.*, 2001, 13, 1–21.
- Leelanandam, C., The anorthosite complexes and Proterozoic Mobile Belt of Peninsular India: a review. *Dev. Precambrian Geol.*, 1990, 8, 409–436.
- Pant, N. C., Kundu, A., Joshi, S., Dey, A., Bhandari, A. and Joshi, A., Chemical dating of monazite – testing of an analytical protocol against independently dated standards. *Indian J. Geosci.*, 2009, 63, 311–318.
- Bhandari, A., Pant, N. C. and Bhowmik, S. K., ~1.6 Ga ultrahigh temperature granulite metamorphism in the Central Indian tectonic zone: insights from metamorphic reaction history, geothermo barometry and monazite chemical ages. *Geol. J.*, 2011, 46, 198–216.
- Bhowmik, S. K., Wilde, S. A., Bhandari, A. and Sarabadhikari, A. B., Zoned monazite and zircon as monitors for the thermal history of granulite terranes: an example from the Central Indian tectonic zone. J. Petrol., 2014, 55(3), 585–621.
- 53. Bowie, S. H. U. and Horne, J. E. T., Cheralite, a new mineral of the monazite group. *Mineral. Mag.*, 1953, **30**, 93–99.
- 54. Taylor, S. R. and McLennan, S. M., *The Continental Crust: Its Composition and Evolution*, Blackwell, Oxford, 1985, p. 318.
- 55. Felsche, J., Yttrium and lanthanides. In *Hand Book of Geochemistry* (ed. Wedephol, K. H.), Springer-Verlag, 1976, vol. II, p. 42.
- Henderson, P. and Pankhrust, R. J., Developments in Geochemistry, Rare Earth Element Geochemistry, Elsevier, Amsterdam, 1984, p. 510.

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- Murata, K. J., Dutra, C. V., Da Costa, M. T. and Branco, J. J. R., Composition of monazite from pegmatites of eastern Minas Gerais, Brazil. *Geochim. Cosmochim. Acta*, 1959, 16, 1–14.
- 58. Deer, W. A., Howie, R. A. and Zussman, J., In *Rock Forming Minerals*, Longman, London, 1975, vol. 5, pp. 339–346.
- 59. Mason, B. and Moore, C. B., *Principles of Geochemistry*, John Wiley, Canada, 1985, 4th edn, pp. 1–344.
- Zhu, X. K. and O'Nions, R. K., Monazite chemical composition: some implications for monazite geochronology. *Contrib. Mineral. Petrol.*, 1999, 137, 351–363.
- 61. Rickers, K., Mezger, K. and Raith, M. M., Evolution of continental crust in the Proterozoic Eastern Ghats Belt, India and new constraints for Rodinia reconstruction: implication from Sm–Nd, Rb–Sr and Pb–Pb isotopes. *Precambrian Res.*, 2001, **112**, 183– 212.
- 62. Yoshida, M., Funaki, M. and Vitange, P. N., Juxtaposition of India–Sri Lanka–Antarctica in Proteorozoic and Gondwana. In *Study of Geological Correlation Between Sri Lanka and Antarctica* (eds Hirod, Y. and Motoyoshi, Y.), 1990, pp. 118–131.

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Forage and security trade-offs by markhor *Capra falconeri* mothers

Riyaz Ahmad^{1,2,*}, Charudutt Mishra^{1,3}, Navinder J. Singh⁴, Rahul Kaul² and Yash Veer Bhatnagar^{1,3}

¹Nature Conservation Foundation, Mysore 3076/5, IV Cross, Gokulam Park, Mysore 570 002, India
²Wildlife Trust of India, Noida 201 301, India
³Snow Leopard Trust, Seattle, USA
⁴Department of Wildlife, Fish and Environmental Studies, Swedish University of Agricultural Sciences, Umeå 90183, Sweden

Food acquisition and security from predators are primary determinants of habitat use in ungulates. There is usually a trade-off in the response of animals to these two factors, influenced by the individual's reproductive state. Females with vulnerable offspring, after parturition, are expected to compromise food acquisition for security. In temperate species such as the markhor *Capra falconeri*, however, the females give birth at a time when nutritious forage begins to

^{*}For correspondence. (e-mail: riyaz@wti.org.in)